

A novel intramolecular defluorinative cyclization approach to the synthesis of difluoromethylated quinazolic acid derivatives

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tert-Butyl 2-(difluoromethyl)-3-benzyl-6-methoxy-3,4-dihydro-4-quinazoline-4-carboxylic ester (**4**) and its related 6-substituted esters, precursors of a new type of cyclic amino acid, were synthesized *via* intramolecular defluorinative cyclization under basic conditions.

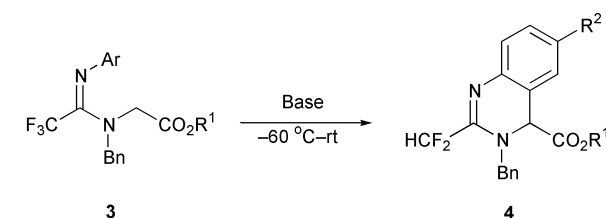
The recent achievement of the synthesis of fluorinated isoserines¹ *via* Wittig rearrangement of **1** to **2** in Scheme 1, engaged us in exploration on the synthesis of other new heteroatom-substituted amino acids and their derivatives which are expected to be candidates for biologically important inhibitors.² Here, we report the discovery of a novel defluorinative cyclization (**3** → **4**) leading to the synthesis of difluoromethylated quinazolic derivatives.



Scheme 1

Difluoromethylated amino acids are known to be important members of the family of fluorinated amino acids.³ Introduction of difluoromethyl groups into nonfluorinated amino acids⁴ or the synthesis of difluoromethyl-substituted amino acids from suitable fluorinated building blocks⁵ is a great challenge in synthetic organofluorine chemistry. Selective defluorination from a trifluoromethyl moiety has been widely employed for promising construction of difluoromethylene moiety due to easy availability of trifluoromethyl compounds.⁶ The present discovery of defluorinative cyclization of trifluoromethyl-substituted iminoamines **3**⁷ provides a novel and efficient approach to the new *N*-heterocyclic *tert*-butyl 2-(difluoromethyl)-3-benzyl-6-methoxy-3,4-dihydro-4-quinazoline-4-carboxylic ester **4**⁸ and the related 6-substituted esters (Scheme 2, Table 1), which is otherwise difficult.

The ester **3** was treated with LTMP (1.3 eq.) in THF at $-60\text{ }^{\circ}\text{C}$ for 10 min or *n*-BuLi (1.2 eq.) at $0\text{ }^{\circ}\text{C}$, and then the temperature was raised to $20\text{ }^{\circ}\text{C}$. The mixture was stirred for several hours (monitored by TLC) or 1 h, respectively.



Ar = 4-MeO-C₆H₄, 4-CH₃-C₆H₄,
3,4-Cl₂-C₆H₃, C₆H₅, Naphthyl
R¹ = *t*-Bu, Me; R² = OMe, Me, H.

Base = LTMP or *n*-BuLi LTMP = Lithium tetramethylpiperidide

Scheme 2

A sterically hindered base (LTMP) provided better yield (63%) of **4a** than LDA (47%). In the 3,4-dichloro-substituted iminoamine (entry 7 in Table 1), *n*-BuLi was found to be more useful than LTMP. Both electron-withdrawing and -donating substituents on the aromatic ring affected the formation of **4**. The 2,6-dimethylphenyl compound (entry 9) afforded no corresponding cyclized compound **4** and the starting substrate was recovered.

The quinazolic esters (**4a–f**) obtained in the reaction are liquids and their structures were elucidated by spectroscopic and elemental analyses, along with X-ray crystallographic structural analysis of its methyl ester (**4g**),⁹ which are colorless needle crystals. Existence of both CHF₂ group and a 1,2,4-trisubstituted benzene ring in **4a** (entry 1 in Table 1, Ar = *p*-methoxyphenyl, R¹ = *t*-Bu) was rationalized by the observation of a doublet at $\delta = 45.8$ ppm ($J_{\text{HF}} = 53.6$ Hz) in ¹⁹F NMR (C₆F₆ as an internal standard) and a triplet at $\delta = 6.33$ ppm ($J_{\text{HF}} = 53.6$ Hz) in ¹H NMR, and three sets of aromatic protons [$\delta = 6.52$ (d, $J = 2.8$ Hz), 6.74 (dd, $J = 8.7$ Hz, 2.8 Hz), and 7.11 ppm (d, $J = 8.7$ Hz)], respectively.

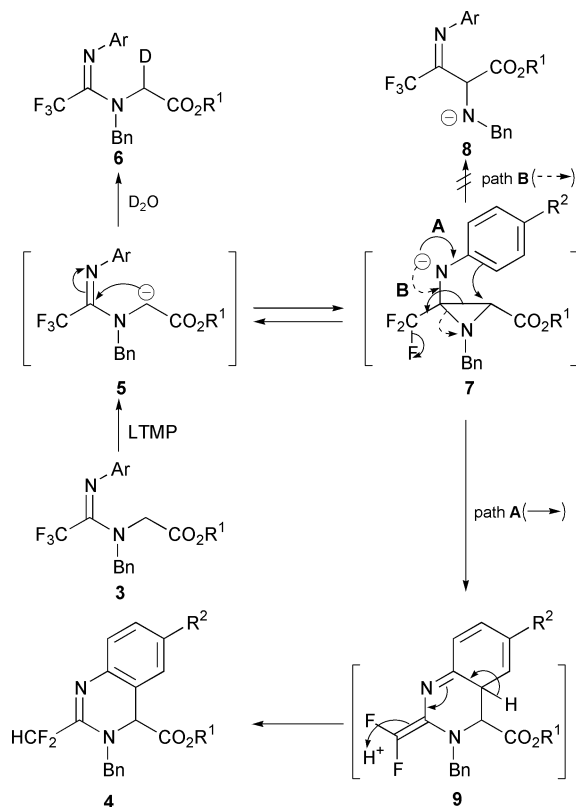
A mechanism of this reaction is tentatively proposed as shown in Scheme 3.

No Stevens-type rearrangement⁹ products **8** were observed although Wittig-type rearrangement is known in fluorinated imino ethers (**1** → **2**).¹ The generation of carbanions **5** by LTMP at lower temperature ($-60\text{ }^{\circ}\text{C}$) was confirmed by the quantitative formation of deuterated product **6** on treating with D₂O. Intramolecular nucleophilic attack of the carbanion onto the imino carbon followed by nucleophilic ring opening of the aziridine type intermediate (**7** → **9**) and simultaneous defluorination would lead to the formation of the quinazoline intermediate **9**, which undergoes base-catalyzed proton migration to the quinazolic products **4**. The rate-determining intramolecular nucleophilic attack *via* intermediate **7** would make the equilibrium (**5** ⇌ **7**) proceed in the forward direction. Thus, the sterically hindered 2,6-dimethyl compound (Table 1, entry 9) was recovered entirely though an α -proton of the carboalkoxy group of **3** (Ar = 2,6-dimethylphenyl) was deuterated by quenching the reaction mixture with D₂O at $-60\text{ }^{\circ}\text{C}$. The preference of path A over path B would arise from the poorer leaving ability of *N*-benzyl anion (Scheme 3) than alkoxide anion in Wittig rearrangement (**1** → **2**).

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Table 1 Yield of **4** in the base-catalyzed rearrangement of **3**

Entry	Ar	R ¹	Base (eq.)	Yield of 4 (%)
1	4-MeOC ₆ H ₄	<i>t</i> -Bu	LTMP (1.3)	4a (63)
2	4-CH ₃ C ₆ H ₄	<i>t</i> -Bu	<i>n</i> -BuLi (1.2)	4b (58)
3	C ₆ H ₅	<i>t</i> -Bu	<i>n</i> -BuLi (1.2)	4c (54)
4	1-Naphthyl	<i>t</i> -Bu	LTMP (1.3)	4d (80)
5	1-Naphthyl	Me	LTMP (1.3)	4e (67)
6	3,4-Cl ₂ C ₆ H ₃	<i>t</i> -Bu	LTMP (1.3)	4f (78)
7	3,4-Cl ₂ C ₆ H ₃	<i>t</i> -Bu	<i>n</i> -BuLi (1.2)	4f (91)
8	3,4-Cl ₂ C ₆ H ₃	Me	LTMP (1.3)	4g (29)
9	2,6-(CH ₃) ₂ C ₆ H ₃	<i>t</i> -Bu	LTMP (1.3)	Recovery of 3



(No 12450356) and the SC-NMR Laboratory of Okayama University for ^{19}F NMR analysis and VBL for X-ray analysis.

Notes and references

- 1 K. Uneyama, J. Hao and H. Amii, *Tetrahedron Lett.*, 1998, **39**, 4079.
- 2 (a) K. L. Kirk and R. Filler, *Recent Advances in the Biomedical Chemistry of Fluorine-containing Compounds*, in *Biochemical Frontiers of Fluorine Chemistry*, ed. I. Ojima, J. R. McCarthy and J. T.

- Welch, ACS Symposium series, Washington, D.C., 1996; (b) J. Kollonitsch, in *Biomedical Aspects of Fluorine Chemistry*, ed. R. Filler and Y. Kobayashi, Kodansha & Elsevier, 1982, pp. 93–122; (c) J. Kollonitsch, L. M. Perkins, A. A. Patchett, G. A. Doldouras, S. Marburg, D. E. Duggan, A. L. Maycock and S. D. Aster, *Nature*, 1978, **274**, 906; (d) E. A. Wang and C. Walsh, *Biochemistry*, 1981, **20**, 7539.
- 3 (a) J. T. Welch and S. Eswarakrishnan, *Fluorine in Bioorganic Chemistry*, John Wiley & Sons, 1994, pp 7–65; (b) V. P. Kukhar and V. A. Soloshonok, *Fluorine-containing Amino Acids, Synthesis and Properties*, John Wiley & Sons, 1994.
- 4 T. Tsushima, S. Ishihara and Y. Fujita, *Tetrahedron Lett.*, 1990, **31**, 3017.
- 5 (a) O. Kitagawa, A. Hashimoto, Y. Kobayashi and T. Taguchi, *Chem. Lett.*, 1990, 1307; (b) J. E. Baldwin, G. P. Lynch and C. J. Schofield, *J. Chem. Soc., Chem. Commun.*, 1991, 736; (c) B. P. Hast and J. K. Coward, *Tetrahedron Lett.*, 1993, **34**, 4917; (d) K. S. Kim and L. Qian, *Tetrahedron Lett.*, 1993, **34**, 7195; (e) T. Tsukamoto, T. Kitazume, J. J. McGuire and J. K. Coward, *J. Med. Chem.*, 1996, **39**, 66.
- 6 (a) J. M. Percy, *Building Block Approaches to Aliphatic Organofluorine Compounds in Topics in Current Chemistry 193*, ed. R. D. Chambers, Springer-Verlag, 1997; (b) P. Cavel, M. P. Legar-lambert, C. Biran, F. Serein-Spirau, M. Bordeau, N. Roques and H. Marzouk, *Synthesis*, 1999, 829; (c) K. Uneyama, G. Mizutani, K. Maeda and T. Kato, *J. Org. Chem.*, 1999, **64**, 6717; (d) H. Amii, T. Kobayashi, Y. Hatamoto and K. Uneyama, *Chem. Commun.*, 1999, 1323.
- 7 The esters **3** were readily prepared by the reaction of *N*-benzyl glycinate with trifluoroacetimidoyl chlorides¹¹ in good to excellent yields.
- 8 The 3,4-dihydroquinazolinone skeleton has been mostly constructed by intramolecular condensation of iminoesters of *o*-aminobenzylamines; (a) A. R. Katritzky, G. Zhang and J. Jiang, *J. Org. Chem.*, 1995, **60**, 7625; (b) M. L. El Efrat, B. Hajjem, H. Zantour and B. Baccar, *Synth. Commun.*, 1996, **26**, 3167.
- 9 $\text{C}_{18}\text{H}_{14}\text{O}_2\text{N}_2\text{F}_2\text{Cl}_2$, $M = 399.38$, $Z = 2$, $D_{\text{calc}} = 1.46$, triclinic, $a = 8.788(6)$, $b = 13.147(7)$, $c = 8.763(4)$ Å, $\alpha = 96.347(2)$, $\beta = 113.782(4)$, $\gamma = 78.837(2)^\circ$, $U = 908.300$ Å³, $T = 288$ K, space group $P\bar{1}$ (no. 2), $Z = 2$, $\mu(\text{Mo-K}\alpha) = 3.91$ cm⁻¹, 2549 reflections measured. The final R was 0.062, R_w was 0.090. CCDC 182/1766. See <http://www.rsc.org/suppdata/cc/b0/b005611/> for crystallographic files in .cif format.
- 10 (a) B. P. Mundy and M. G. Ellerd, in *Name Reactions and Reagents in Organic Synthesis*, Wiley, New York, 1988, pp 202–203; (b) T. Thomson and T. S. Stevens, *J. Chem. Soc.*, 1932, 55; (c) I. Coldham, M. L. Middleton and P. L. Taylor, *J. Chem. Soc., Perkin Trans. 1*, 1998, 2817; (d) I. E. Marko, *Comp. Org. Syn.*, 1991, **3**, 913.
- 11 K. Tamura, H. Mizukami, K. Maeda, H. Watanabe and K. Uneyama, *J. Org. Chem.*, 1993, **58**, 32.